

the usual isomerization of a 4-methoxyfuranoquinoline derivative into a N-methylfurano-4-quinolone compound (II).

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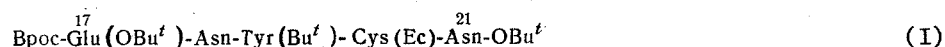
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#### SYNTHESIS OF A SUBSTITUTED FRAGMENT A<sup>17-21</sup> OF HUMAN INSULIN

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In connection with the development of a new scheme for the block synthesis of the A-chain of human insulin, we have performed the synthesis of the substituted pentapeptide (I) corresponding to fragment A<sup>17-21</sup> of human insulin



The pentapeptide (I) was synthesized by a 4 + 1 scheme using as starting materials the pentafluorophenyl ester of N-biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamic acid (II), the pentafluorophenyl ester of N <sup>$\alpha$</sup> -benzyloxycarbonyl-L-asparagine (III), the methylester of N-benzyloxycarbonyl-O-tert-butyl-L-tyrosine (IV), S-ethylcarbamoyl-L-cysteine (V), and the tert-butyl ester of N <sup>$\alpha$</sup> -benzyloxycarbonyl-L-asparagine (VI).

Intermediate compounds were the methyl ester of O-tert-butyl-L-tyrosine (VII), the methyl ester of N <sup>$\alpha$</sup> -benzyloxycarbonyl-L-asparaginyl-O-tert-butyl-L-tyrosine (VIII), the methyl ester of L-asparaginyl-O-tert-butyl-L-tyrosine (IX), the methyl ester of N-biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamyl-L-asparaginyl-O-tert-butyl-L-tyrosine (X), N-biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamyl-L-asparaginyl-O-tert-butyl-L-tyrosine (XI), the pentafluorophenyl of N-biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamyl-L-asparaginyl-O-tert-butyl-L-tyrosine (XII), N-biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamyl-L-asparaginyl-O-tert-butyl-O-tyrosyl-S-ethylcarbamoyl-L-cysteine (XIII), the tert-butyl ester of L-asparagine (XIV), and the pentafluorophenyl ester of N-biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamyl-L-asparaginyl-O-tert-butyl-L-tyrosyl-S-ethylcarbamoyl-L-cysteine (XV).

For the synthesis of the S-ethylcarbamoyl-L-cysteine we used the method suggested by Guttmann [1]. The pentafluorophenyl esters of N-substituted amino acids and peptides were obtained by the method of Kisfaludy et al. [2].

The protected pentapeptide (I) was purified by recrystallization from dioxane-ethyl acetate (1:10). The structure of compound (I) was determined unambiguously by the scheme of synthesis, and its individuality was checked by chromatography and the results of analytical determinations.

tert-Butyl Ester of N-Biphenylisopropoxycarbonyl- $\gamma$ -tert-butyl-L-glutamyl-L-asparaginyl-O-tert-butyl-L-tyrosyl-S-ethylcarbamoyl-L-cysteinyl-L-asparagine (I). mp 192-193°C (decomp.),  $[\alpha]_D^{25} - 38.6^\circ$  (c, 1.0; dimethylformamide). TLC on Silufol UV-254 plates; R<sub>f</sub> 0.30 (methanol-chloroform (1:9)), 0.52 (methanol-chloroform (2:17)). Found, %: C 60.49; H 7.05; N 9.87. C<sub>53</sub>H<sub>79</sub>N<sub>8</sub>O<sub>14</sub>S. Calculated, %: C 60.33; H 7.10; N 10.00. Amino acid analysis (the cysteine was not determined): Glu 1.00, Asp 1.96, Tyr 0.87.

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NOMOGRAM FOR THE RAPID DETERMINATION OF THE CONTRIBUTION  
OF SCATTERED LIGHT TO THE ABSORPTION SPECTRA OF BIOPOLYMER  
SOLUTIONS

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It is known that in the use of the absorption spectra of solutions of biopolymers (proteins, nucleic acids) in the ultraviolet region, in addition to the true absorption  $D_{\text{true}}$  due to the chromophores an additional optical density  $D_{\text{sc}}$  due to the Rayleigh scattering of the light is recorded, and

$$D_{\text{true}} = D_0 - D_{\text{sc}} \quad (1)$$

where  $D_0$  is the measured optical density.

It must be noted that  $D_{\text{sc}}$  may amount to 80% of  $D_0$  [1], and therefore the measurement of  $D_{\text{sc}}$  is of prime importance.

At the present time it is customary to find  $D_{\text{sc}}$  by means of the relation

$$\lg D_{\text{sc}} = \lg a - n \lg \lambda, \quad (2)$$

which follows from the generalized formula of Rayleigh scattering [2]. However, it is extremely laborious and requires the expenditure of much time to find  $D_{\text{sc}}$  with the aid of formula (2).

The nomogram shown in Fig. 1 permits the process of finding  $D_{\text{sc}}$  to be considerably shortened and simplified. For this purpose it is sufficient to transfer the absorption spectrum pointwise to a nomogram, to find the region of wavelengths in which the points fall on a straight line (this region will correspond to the absence of true absorption), and to perform linear extrapolation. The value of  $D_{\text{sc}}$  must be found by recalculation from the extrapolation line with the ordinate corresponding to the selected wavelength. The resolving capacity of the nomogram is 0.2-0.3% of the optical density of the scattered light to be determined. The accuracy of measuring  $D_{\text{sc}}$  is determined by the accuracy of plotting the experimental points on the nomogram and the accuracy of extrapolation. The time necessary for determining  $D_{\text{sc}}$  is about 1 minute.

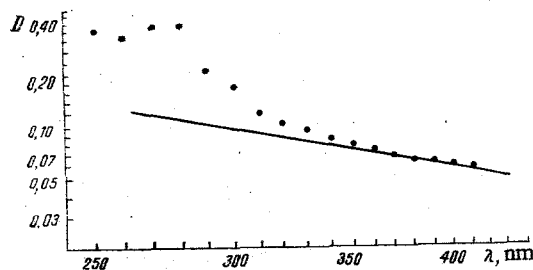


Fig. 1. Nomogram for determining the contribution of the apparent optical density due to the scattering of light in a measured absorption spectrum. The points show the absorption spectrum of an aqueous solution of thyroglobulin. The optical density of light scattering at 280 nm is 0.110.

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